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Title: Determination of triamcinolone acetonide in silicone oil and aqueous humor of vitrectomized rabbits' eyes: Application for a pharmacokinetic study with intravitreal triamcinolone acetonide injections (Kenalog® 40)

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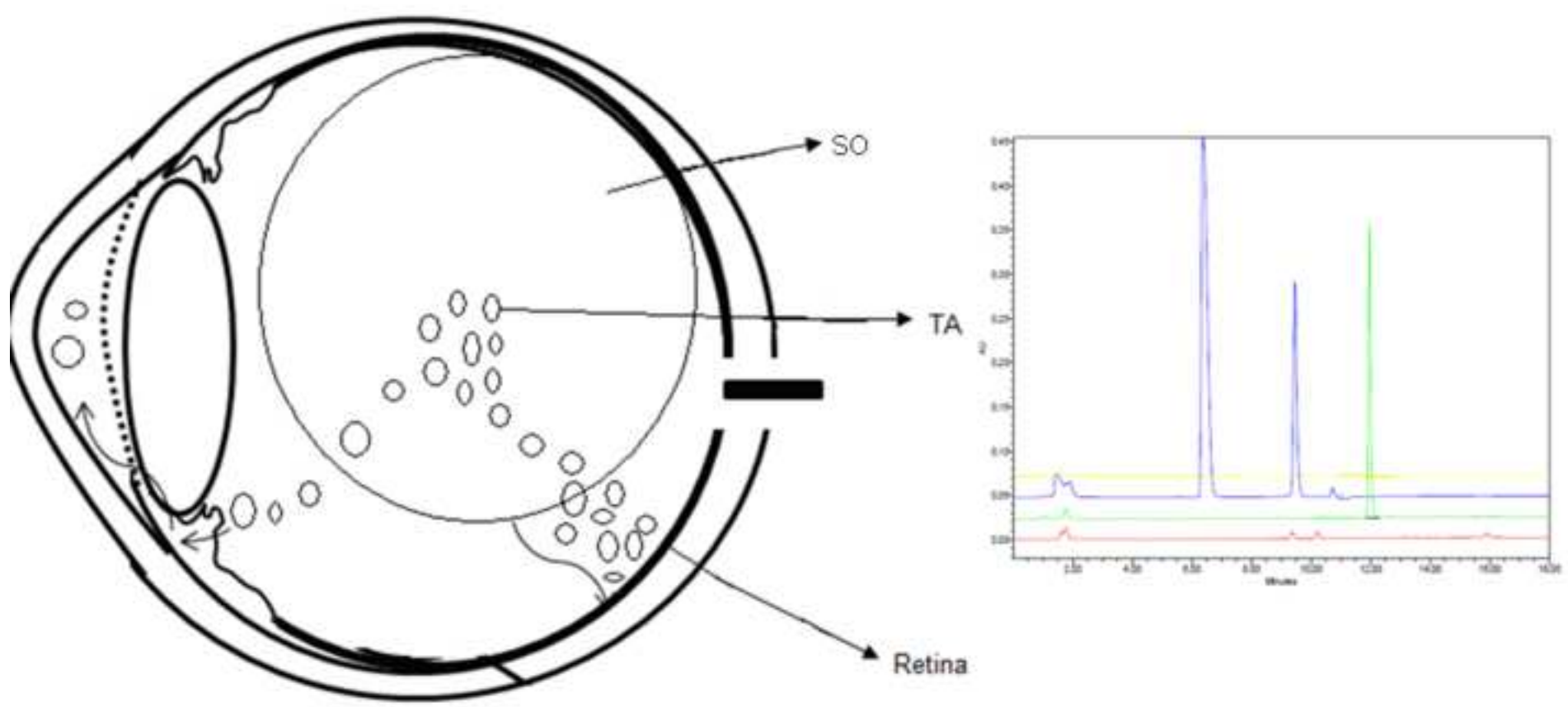
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- High-performance liquid chromatography method was developed and validated.
- Administration of TA injections in rabbits' eyes.
- Method was successfully applied to quantify the drug in SO and aqueous humor.
- TA remained in SO and aqueous humor of rabbits' eyes for 4 weeks.
- SO may play an important role in the elimination of lipophilic drugs.

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1 Determination of triamcinolone acetonide in silicone oil and aqueous humor of
2 vitrectomized rabbits' eyes: Application for a pharmacokinetic study with intravitreal
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4

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17

18 **ABSTRACT**

19 A simple and accurate method including liquid-liquid extraction and protein precipitation
20 procedures from silicone oil and aqueous humor samples followed by high-
21 performance liquid chromatography (HPLC-UV) was developed and validated to
22 determine the pharmacokinetic profile of triamcinolone acetonide in silicone oil and
23 aqueous humor of rabbits' eyes submitted to the *pars plana* vitrectomy surgery. The
24 method was successfully applied to quantify the drug remaining in silicone oil and
25 aqueous humor (LOQ range of 1µg/mL). The triamcinolone acetonide remained in

26 silicone oil and aqueous humor of vitrectomized rabbits' eyes for four weeks after the
27 intravitreal injections.

28

29 Keywords: Silicone oil, Triamcinolone acetonide, Intravitreal injections, Vitrectomy,
30 HPLC-UV.

31

32 **1. Introduction**

33 The *Pars Plana* vitrectomy surgery, in which the humor vitreous is replaced with
34 an internal tamponade, the silicone oil (SO), is performed in order to repair the retinal
35 detachment caused by proliferative vitreoretinal diseases [1,2]. The intravitreal
36 injections of triamcinolone acetonide (TA) (Fig. 1), a synthetic lipophilic corticosteroid
37 with low solubility in aqueous solution, are applied to overcome surgical complications
38 [3-5]. However, the quantity of the drug present in the SO after vitrectomy surgery is
39 unknown, possibly leading to inadequate therapy.

40

INSERT FIGURE 1

41 Some methods have been reported for determining TA in rabbit and human's
42 eyes. Oliveira et al determined TA concentration in humor vitreous of rabbits' eyes by
43 HPLC-UV using a C-18 column and ACN/H₂O (60:40 v/v) as mobile phase [6].
44 However, this method was applied to quantify TA only in humor vitreous therefore
45 modifications need to be done to quantify the drug in different matrices. In other study,
46 Beer et al applied a HPLC-MS method to determinate TA in aqueous humor of human's
47 eyes [7]. Regardless the high sensitivity of HPLC-MS method, the major limitation of
48 this kind of analysis is the matrix effects in which the matrix coextracted with the
49 analyte can alter the signal response. Furthermore, the methods using HPLC-MS are
50 expensive and not readily available in all laboratories [8]. In this study, we chose to
51 develop a HPLC-UV method to quantify TA in SO and aqueous humor of rabbit's eyes,

52 since it is simple to perform, it is cost and time-effective and have low limits of
53 detection.

54 Thus, in the present work an easy, manageable and rapid HPLC-UV method
55 combined with SO drug extraction and aqueous humor protein precipitation for
56 quantifying TA in vitrectomized rabbits' eyes was developed and validated. The method
57 was applied for *in vivo* study in which rabbits groups received injections of TA.

58

59 **2. Experimental**

60 *2.1 Chemical and reagents*

61 TA reference standard was purchased from the Sigma-Aldrich Co. (St. Louis,
62 MO, USA). SO (5000 cTs) was obtained from Ophthalmos (São Paulo, SP, Brazil). TA
63 aqueous suspension was purchased from (Bristol-Myers, NJ, USA). Ultra-pure water
64 was obtained from a Millipore system (Bedford, MA, USA). Acetonitrile and methanol
65 (HPLC grade) were purchased from Tedia (Fairfield, OH, USA) and ethyl acetate
66 (HPLC grade) was obtained from Vetec (Rio de Janeiro, RJ, Brazil).

67 *2.2 Instrumentation and analytical conditions*

68 The reversed-phase HPLC system was a Waters apparatus (Massachusetts,
69 USA) equipped with a 717 plus autosampler model, consisting of a 515 pump and a
70 486 ultraviolet detector. Data collection and integration were achieved using Enpower
71 (version 6.2) software. The analytical column used was a Chromolith® Merck C₁₈ (5 µm
72 particle size; 100 x 4.6 mm i.d). The mobile phase used for TA quantification consisted
73 of HPLC water (A) and acetonitrile (B). Separation was carried out applying a gradient
74 elution at ambient temperature using a flow rate 1.0 mL/min. The mobile phase
75 combinations were: 0 min, 90% A, 10% B; 10 min, 90→50% A, 10→50% B; 5 min of
76 isocratic elution; 50% A, 50% B; 3 min, 50→90% A, 50→10% B. A re-equilibration

77 interval of 15 min in the initial conditions was introduced between subsequent
78 analyses. Detection was achieved at 239 nm.

79 *2.3 Preparation of standard solution*

80 Stock solution of TA was prepared by dissolving the accurately weighed
81 reference substance in methanol. The working solution of TA was prepared
82 immediately before the use by diluting the stock solution to a final concentration of 40
83 $\mu\text{g/mL}$.

84 *2.4 Extraction of TA from SO*

85 First, 300 μL of ethyl acetate were added in a plastic tube containing 100 μL of
86 SO. This solution was mixed in a vortex for 5 minutes. Then, an aliquot (1000 μL) of TA
87 working solution was transferred to the plastic tube and mixed in a vortex for 5 minutes.
88 After that, the final solution was evaporated during 48 hours. The extraction consisted
89 of the addition of different solvents such as acetonitrile, methanol and water to 100 μL
90 of evaporated SO. Different extraction times were analyzed (5, 10, 15 min) as well as
91 different quantity of solvents (500 μL two times, and 1000 μL). The mixture was stirred
92 for 5 minutes and centrifuged at 300 x g for 5 minutes. The supernatant was then
93 collected, filtered and transferred to a vial. A 20 μL aliquot was injected into the
94 chromatographic system.

95 *2.5 Samples preparation*

96 A 500 μL aliquot of acetonitrile was added to the TA SO and aqueous humor
97 samples. The samples were vortex mixed for 5 minutes and centrifuged at 300 x g for 5
98 minutes at ambient temperature. The supernatant was collected, filtered and
99 lyophilized. The residue from lyophilization was resuspended in 100 μL of acetonitrile.
100 A 20 μL aliquot was injected into the HPLC system to determinate the amount of TA.

101 *2.6 TA method validation*

102 The validation process was carried out as described in the literature [9]. TA
103 method selectivity was assayed by injection of SO and aqueous humor blank extracted
104 samples and by the injection of TA blank injectable suspension. TA method linearity
105 was assessed by six and five-point calibration curves in methanol in triplicate in three
106 consecutive days. The concentration range evaluated was 1.0-120.0 $\mu\text{g/mL}$. The
107 curves were evaluated by residuals and fitted by weighted linear regression. The LOQ
108 was established analyzing the means of six replicates. The LOD was defined as the
109 concentration giving a sign-to-noise ratio (S/N) of 3. The intra-day precision was
110 evaluated by RSD values of sample solutions analyzed on the same day ($n=6$), at the
111 middle concentration of the calibration curves, whereas inter-day precision was
112 accessed by analyzing sample solutions prepared on two different days, by two
113 different analysts ($n=12$). The extraction recovery of the method was determined by
114 comparing the peak areas obtained from the SO samples with those of direct injected
115 standards, at the same concentration. The evaluation was done by analyzing five
116 replicates containing 1, 40, 120 $\mu\text{g/mL}$ of TA. The stability of the analyte in SO was
117 evaluated using the working solution in six replicates. The drug was left in ambient
118 temperature for 72 hours in SO. The stability of the drug during the run-time in the
119 HPLC auto-injector was investigated of one concentration levels 40 $\mu\text{g/mL}$. Samples
120 were prepared and kept in the sample rack of the auto-injector and injected into the
121 HPLC system 24h after preparation. Then the stability was investigated by analyzing
122 the concentrations found.

123 *2.7 Application to a pharmacokinetic study*

124 The validated method was used to determine the concentration of TA in rabbits'
125 vitrectomized eyes after administration of the intravitreal injection of TA. Nine New
126 Zealand albino adult male rabbits weighing 2.0-2.5 Kg were used for *in vivo* studies. All
127 experimental procedures involving animals were performed in agreement with the

128 Ethics Committee of Universidade de São Paulo (USP-Ribeirão Preto) and according
129 to the Association for Research in Vision and Ophthalmology (ARVO) statement for the
130 use of animals in ophthalmic and vision research. They were placed in two groups;
131 group I (n=3) underwent standard *pars plana* vitrectomy with the injection of 1000 μL of
132 SO and served as experiment control; group II (n=6) underwent standard *pars plana*
133 vitrectomy with the injection of 1000 μl of SO and 100 μl of TA (Kenalog[®] 40; 40 mg/mL)
134 were injected intravitreally just after the surgery was over. In group I and II SO and
135 aqueous humor were collected from vitreous cavity and anterior cavity at 1 and 4
136 weeks after the administration of the drug by intravitreally injection. The samples were
137 immediately placed in eppendorf tubes and stored at -20°C until HPLC-UV analysis.

138

139 **3. Results and discussion**

140 *3.1 Conditions for HPLC-UV*

141 The selectivity aspect was accessed by analyzing whether blank solutions
142 interfere at TA retention time. No interfering peaks were detected in the analyte peak
143 region (Fig. 2). The retention time of TA was approximately 12.3.

144

INSERT FIGURE 2

145 To date, the exact TA amount present in the vitreous cavity after vitrectomy
146 surgery is not known. In this study we presented a combination of a HPLC-UV method
147 with SO extraction which provides an exact amount of the drug in the vitreous cavity.

148 *3.2 Development of the procedure for sample extraction*

149 We have chosen the ethyl acetate as the ideal solvent once it was able to
150 solubilize both constituents in order to obtain a dispersed system. The extraction
151 studies were conducted by testing three solvents (acetonitrile, methanol and water).
152 The solvent that provided highly efficient separations, a good symmetric peak and
153 recovery rates higher than 90% was acetonitrile. The best recovery was achieved with

154 two aliquots of 500 μL , which can be explained based on the greater contact surface
155 between the drug and the solvent. No significant differences were found between the
156 different times therefore the 5 min time was chosen because it was the shortest time.
157 The recovery method here developed is simple, robust and efficient, resulting in a fast
158 and easily-handled analysis.

159 *3.3 Method validation*

160 No significant interference was detected in the retention times of the analyte in
161 the chromatograms (Fig. 2). Calibration curves were shown to be linear over the range
162 1.0-120.0 $\mu\text{g/mL}$. Typical standard curve was $y=4.05 \times 10^{-4}x + 4.33 \times 10^{-4}$, with a
163 weighted factor $1/x$. Regression coefficient was 0.9992, showing an excellent
164 correlation. Linearity data are presented in Table 1. The obtained LOQ and LOD were
165 respectively 1.0 $\mu\text{g/mL}$ and 0.2 $\mu\text{g/mL}$. The obtained data for intra-run and inter-run
166 precision and accuracy are shown in Table 2. The mean R.S.D. values in the intra-run
167 precision was 2.17% and inter-run precision was 1.22% and. The mean accuracy value
168 in the intra-run assay was 92.0% and in the inter-run assay was 85.5%. The mean
169 recovery rates of TA (n=18), determined at three concentrations, was 101% (Table 3).
170 The results of stability experiments showed that there were no significant degradation
171 of TA in SO samples following 72 hours after extraction and no significance
172 decomposition was observed after the reconstituted samples of TA had been stored in
173 auto injector at room temperature for 24 h. The measure concentration after 24h and
174 72h were all $> 100\%$ of the initial values for the drug at the concentration of 40 $\mu\text{g/mL}$.

175 ***INSERT TABLE 1***

176 ***INSERT TABLE 2***

177 ***INSERT TABLE 3***

178 *3.4 Application to a pharmacokinetic study*

179 The HPLC method was successfully applied to determine the TA in a
180 pharmacokinetic study in rabbits' vitrectomized eyes. Fig. 3 shows the concentrations
181 of intravitreal TA in SO and aqueous humor (group II). The mean concentration of
182 intravitreal TA that remained in SO was of $60.76 \pm 1.6 \mu\text{g/mL}$ (0.15% of initial injection
183 concentration) and $28.20 \pm 8.7 \mu\text{g/mL}$ (0.07%), and in aqueous humor it was of $3.64 \pm$
184 $1.6 \mu\text{g/mL}$ and $7.35 \pm 0.9 \mu\text{g/mL}$ on weeks 1 and 4 respectively. In this study, we
185 showed the remaining TA concentration, both in aqueous humor and in SO in vitreous
186 cavity, which can provide additional information to develop a more reliable TA
187 injections intervals and initial dosage. Additionally, we obtained a TA half-life values of
188 3.5 days in the vitreous cavity of vitrectomized rabbits' eyes whereas in other study it
189 was found a higher TA half-life in non-vitrectomized rabbits' eyes ($8 \pm 2,8$ days) after
190 an initial dosage of 4mg/mL [6]. Our results have lead us to affirm that the TA is
191 cleared 2.3 times faster than in non-vitrectomized eye cavity, suggesting that the SO
192 plays an important role in the TA clearance kinetic profile.

193 **INSERT FIGURE 3**

194 **4. Conclusion**

195 To our knowledge, this is the first method for quantifying TA in SO using HPLC-
196 UV. The method was validated and showed to be robust and reproducible allowing
197 elucidating the *in vivo* pharmacokinetic of TA in vitrectomized eyes. We successfully
198 showed that the SO may play an important role in the elimination of lipophilic drugs
199 which is additional information when planning intravitreal injection of TA in
200 vitrectomized eyes. In this work, we suggest that the SO potentially can interfere in the
201 elimination of lipophilic drug administrated in vitreous cavity.

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207 **Competing Interests**

208 The authors declare that they have no competing interests.

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Table 1

Precision and accuracy data of back calculated concentrations of calibrations samples for triamcinolone acetonide in organic solvent.

Analyte	Nominal concentrations($\mu\text{g}/\text{mL}$)	Observed concentration ($\mu\text{g}/\text{mL}$)	Precision (%R.S.D.)	Accuracy (%)
Triamcinolone acetonide	1	1.1	2.6	109.0
	10	10.1	4.0	101.1
	20	20.5	2.3	102.3
	40	41.8	4.1	104.6
	80	78.3	1.6	97.9
	120	120.4	3.9	100.3

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Table 2

Precision and accuracy data for triamcinolone acetonide in organic solvent by HPLC-UV.

Analyte	Assay	Nominal concentration ($\mu\text{g/mL}$)	Observed concentration ($\mu\text{g/mL}$, mean \pm S.D.)	Precision (%R.S.D.)	Accuracy (%)
Triamcinolone acetonide	Intra-run (n=6)	40.0	36.8 \pm 0.8	2.8	92.0
	Inter-run (n=12)	40.0	34.2 \pm 0.7	2.2	85.5

S.D., standard deviation; R.S.D., relative standard deviation

Table 3

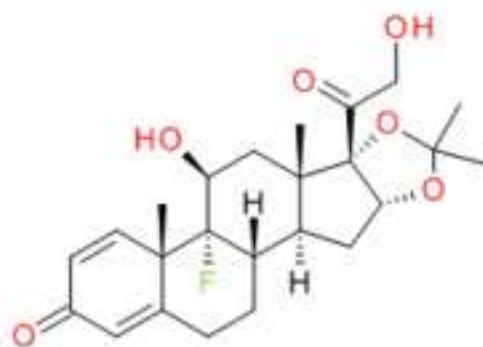
Recovery data for triamcinolone acetonide from silicone oil by HPLC-UV.

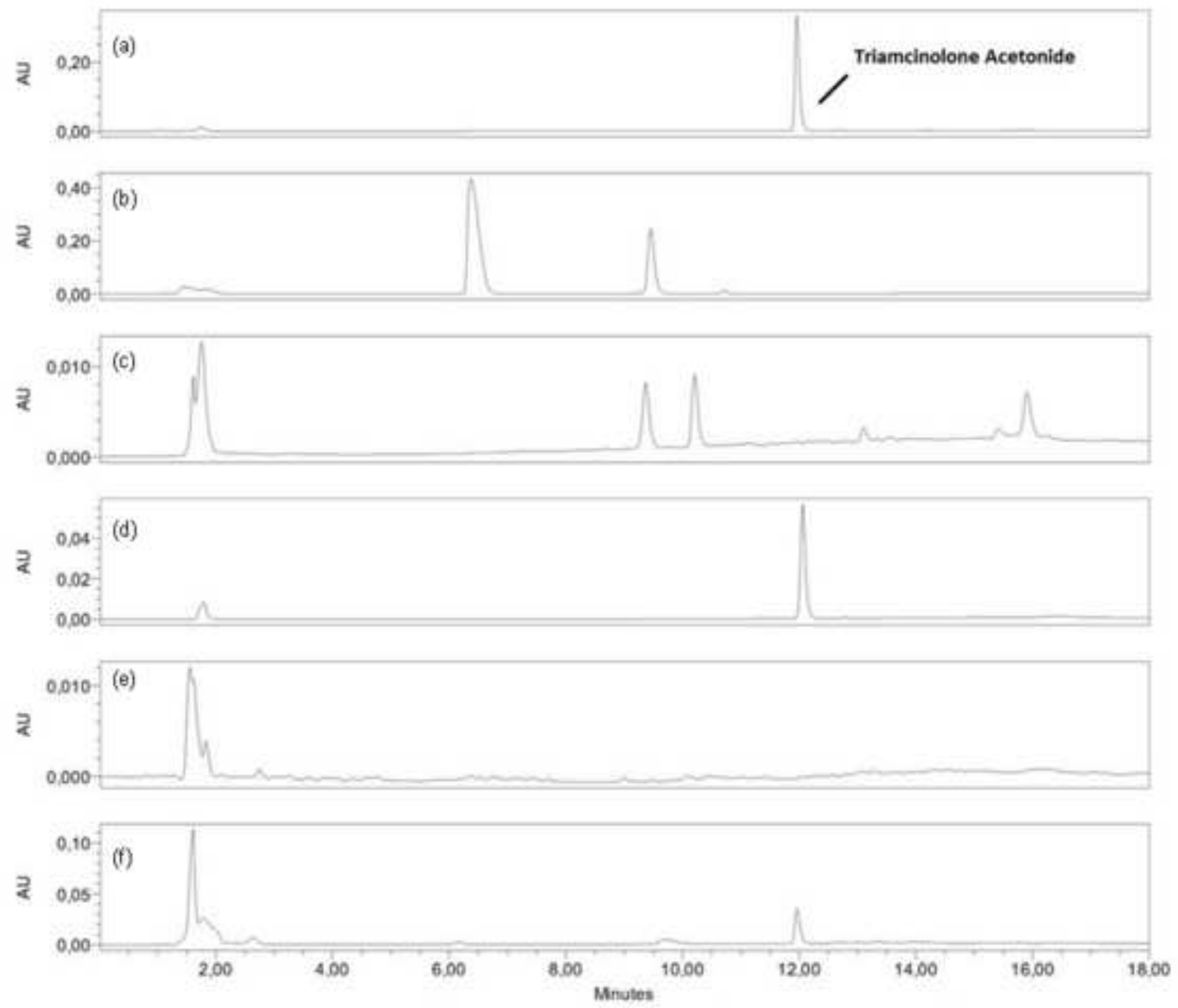
Analyte	Nominal concentration ($\mu\text{g/mL}$)	Recovery (%)	%R.S.D.	Mean recovery (%)
Triamcinolone acetonide ($n=9$)	1.0	97.9	2.1	101.0
	20	105.1	1.8	
	80	99.9	3.8	

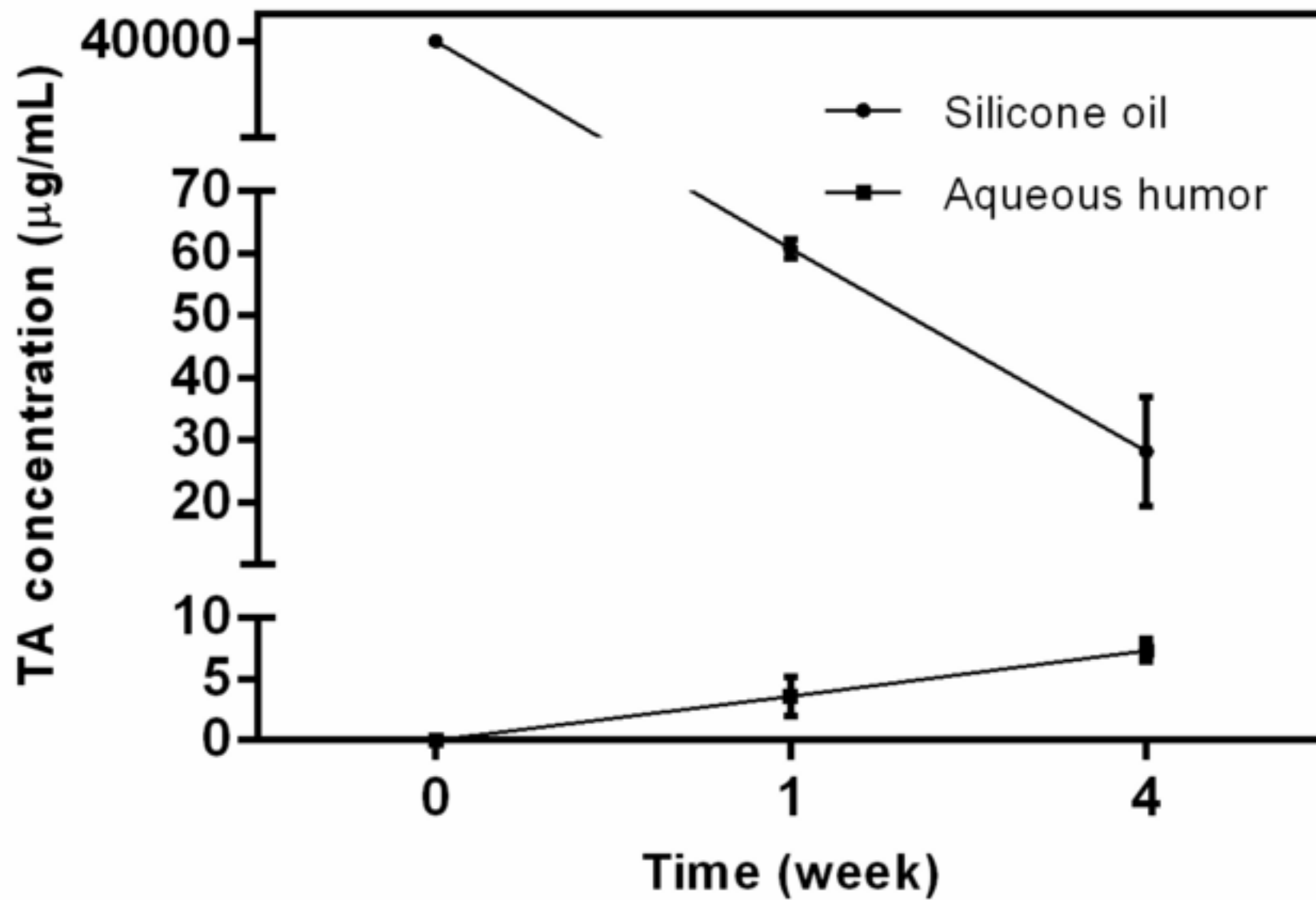
R.S.D., relative standard deviation.

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FIGURES CAPTION

Fig. 1. Chemical structure of triamcinolone acetonide.

Fig. 2. Representative chromatogram of: (a) triamcinolone acetonide standard (b) triamcinolone acetonide blank formulation (c) extraction of silicone oil blank sample (d) extraction of triamcinolone acetonide from silicone oil sample (e) aqueous humor blank sample (f) triamcinolone acetonide from aqueous humor sample. The chromatograms demonstrated no interfering peaks co-eluted with the compound of interest.

Fig. 3. Triamcinolone acetonide profile elimination after an intravitreal injection of 40mg/mL during the follow-up 4 weeks. TA concentrations ($\mu\text{g/mL}$) are shown in silicone oil and aqueous humor. The values are shown as mean \pm standard deviation, $n = 3$.

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